

# Graphene-Oxide and Hydrogel Coated FBG-Based pH Sensor for Biomedical Applications <sup>†</sup>

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**Abstract:** A hydrogel coated fibre grating-based pH sensor for biomedical applications has been realised, where Graphene Oxide (GO) had been used to enhance the bonding between the coating and the fibre. Two methods of deposition of GO were analysed i.e., evaporation and co-electroplating. The paper concludes that the system of GO evaporated on the fibre + the hydrogel has a sensitivity much higher,  $(6.1 \pm 0.5)$  pm/pH, than the system of Cu and GO co-electroplated + the hydrogel,  $(1.9 \pm 0.1)$  pm/pH, for a pH range between 2 to 10. The other conclusion is that the first system has a less coating bonding energy with the optical fibre whereas the second system has a stronger bonding energy, with better durability.

**Keywords:** graphene oxide; fibre optic; pH; sensing; FBG; roughness; surface functionalisation; wettability; GO; biomedical application

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## 1. Introduction

In living organisms, metabolic processes can only take place if specific physical and chemical conditions are met and this is the reason for maintaining the pH of extracellular fluids, including blood plasma, at 7.4 by efficient buffering systems [1]. The level of pH in stomach content could lead to the identification of various illnesses, especially in children and thus there had been a growing interest in research on non-invasive and intrinsically safe sensing techniques for pH measurement. Traditional pH sensing techniques use acid-based indications and electrodes, while in comparison, fibre optic-based pH sensors have the advantages of electrical inertness, immunity to electromagnetic interference, and miniature size. This makes them ideal for biomedical applications, especially considering the volume of information and safety aspects. Fibre optic-based pH sensor techniques are of two-fold: utilization of absorption and fluorescence; coating pH-sensitive material on fibre [2–5].

The concept of in fibre grating devices for pH sensing includes the fibre grating, i.e., Fibre Bragg Grating (FBG), acting as the basis of the device and the pH sensing mechanism exploits the strain effect induced on the FBG through the swelling of a pH-sensitive polymer coating applied on the grating region.

Shivananju et al. had proven [6] that hydrogel made by alginic acid sodium salt (2% in aqueous solution at 20 °C,  $C_6H_9NaO_7$ ) gets a form of viscous hydrocolloid or gels in water by crosslinking with the linker  $CaCl_2$  (0.8 M). The crosslinking agent of the divalent cations  $Ca^{2+}$  in the polymeric network form the hydrogel by electrostatic interaction with the carboxyl groups of sodium alginate, which are sensitive to pH variation. The sodium alginate has been selected because of its minimal invasive properties (biocompatibility and ease of gelation) and for its vast application in biomedical science and engineering [7].

Hydrogels offer some limitations when they are coated on substrates due to the low interfacial toughness. For instance, they have a very low level of coefficient of friction of gels  $\mu \sim 10^{-3}$ – $10^{-4}$  [8], while bonding a water-rich hydrogel [9]. Some studies used either approaches like functionalising the substrate surface with some nonporous solids either via self-assembled monolayers [10], or using the strong bonding between the hydrogel and elastomers by employing the ultraviolet-initiated polymerisation [11] or adding nanoparticles of silica, acting as connectors between polymer chains [12]. In this work, the friction has been done by increasing the roughness and hence increasing the wettability of the hydrogel on the substrate material (fibre optic) by using graphene oxide.

## 2. Materials and Methods

Graphite with the particle size less than 50  $\mu\text{m}$  was chosen to produce and lower the degradation problem of Graphene Oxide [13]. GO was made by applying modifications to the method described by Zaaba et al. [14], synthesized using the Hummers modified method. The FBG was first coated with GO using two different GO deposition techniques (i.e., evaporation and co-electroplating), to increase the wettability of the hydrogel, and re-coated with the polymeric layer containing sodium alginate and calcium chloride. While doing so, the Bragg wavelength was continuously monitored with the interrogator system Micron Optics SM125 and then 24 h were allowed for the sensor to stabilise in air at room temperature. The applied method is detailed herewithbelow.

### 2.1. Coating Procedure

#### 2.1.1. Evaporation Deposition Method

A solution of 1 g/L of GO (pH  $2.5 \pm 0.1$ ) in de-ionised water was prepared and then subjected to an ultrasonic bath for one hour to exfoliate the GO (i.e., in order to avoid conversion to graphite oxide [15]). The FBG sensors were then put onto a petri dish, 1 mL of GO solution (1 g/L) was added on top of the fibre. The oven was set at 90 °C for 20 min to allow evaporation of water from the solution. Then the fibre was turned over to the other side and the same coating process was repeated. Due to the evaporation method, at 90 °C, the concentration of GO solution gradually became more concentrated and more attached to the fibre. The sample was analysed using the SEM (Tescan Vega 3 LMU).

#### 2.1.2. Co-Electroplating Deposition Method

The technique used involved co-electroplating of Cu and the GO simultaneously onto the cathode while placing the optical fibre on top of the cathode. The copper plates used were of ( $0.50 \pm 0.01$ ) mm depth, ( $10.00 \pm 0.01$ ) mm width and ( $40.00 \pm 0.01$ ) mm length. The GO solution was of 1 g/L (pH  $2.5 \pm 0.1$ ) and was left in the ultrasonic bath again for one hour before the deposition process begun. The fibre was placed onto the cathode with a potential of 15 V for 10 min and a distance between the two electrodes of 2 cm. The fibre was then placed in the oven for 20 min at 90 °C and the process was repeated again by turning the fibre over to the other side and analysed using the SEM as discussed earlier.

#### 2.1.3. Hydrogel Coating

The hydrogel layer deposition was achieved using a dip-coating mechanism in which the fibre (containing GO coated FBG) was immersed (for 2 min) into the solution of sodium alginate and then extracted from the solution at a speed of 5 mm/min, since the coating was neither uniform nor continuous at higher coating speeds. The fibre was then immersed into the calcium chloride solution for 10 s and extracted at a speed of 60 mm/min (since the hydrogel could contain more  $\text{Ca}^{2+}$  if it is immersed for longer period). Before doing the pH analysis, the two systems were left to dry for 24 h.

## 3. Results and Discussion

Coating of hydrogel on fibre could be challenging due to poor bonding between the polymer and the glass surface, thus increasing the surface area by increasing the roughness of the surface to

have better wettability. In this study, both processes of deposition of GO had been investigated, followed by the coating of the hydrogel. It could be seen that a non-uniform and non-continuous coating for both the two methods of deposition had been achieved, increasing the surface roughness, as shown in Figure 1. The electrostatic coating given by the evaporation had less bonding strength between the particles and the optic fibre, as shown in Figure 1a. The hydrogel was then applied as discussed in Section 2.1.3.

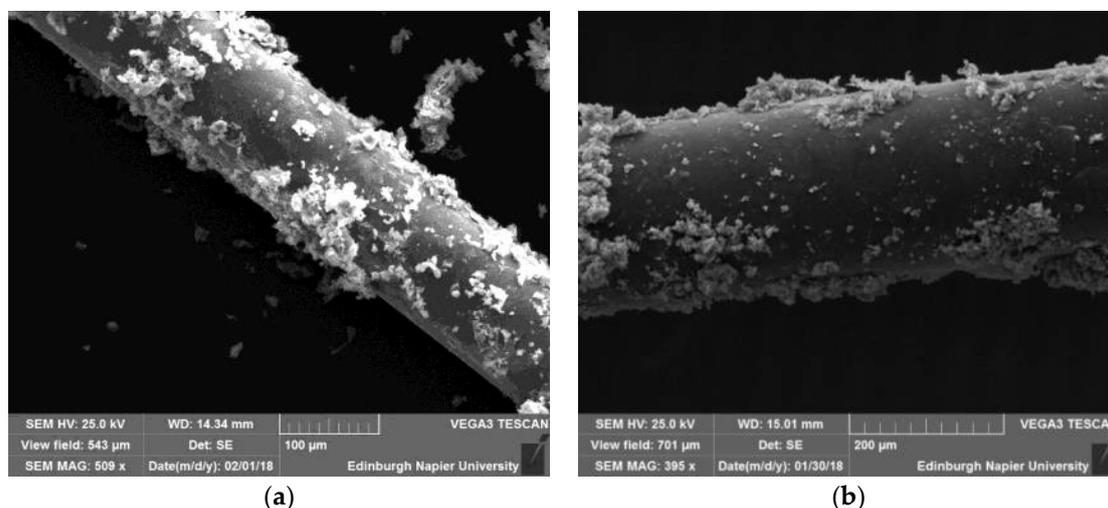


Figure 1. SEM image of the fibre with (a) GO evaporated; (b) GO co-electroplated.

It should be noted while producing the sensor for biological applications, that hydrogel degrades under different mechanisms depending on pH. When it is subjected to pH above 10.0, it decomposes because of  $\beta$ -elimination mechanisms. Below pH 5.0 acid-catalysed hydrolysis causes hydrogel degradation [16]. For this reason, the system was immersed for 2 min in different pH solutions. There was a correspondence between the variation in pH and the respective variation in the FBG resonance wavelength for both GO deposition methods at room temperature, as depicted in Figure 2. There was a dramatic difference in sensitivity across the two methods. The evaporated GO with the hydrogel having a sensitivity of  $(6.1 \pm 0.5)$  pm/pH whereas the co-electrodeposited GO with the hydrogel has a relatively less sensitivity of  $(1.9 \pm 0.1)$  pm/pH. Even though the sensitivity is higher in the first case, the coating was not strongly bonded to the fibre. In the second method, even though with less sensitivity, the overall performance was higher in terms of strong bonding to the optic fibre due to the strongly attached GO particles.

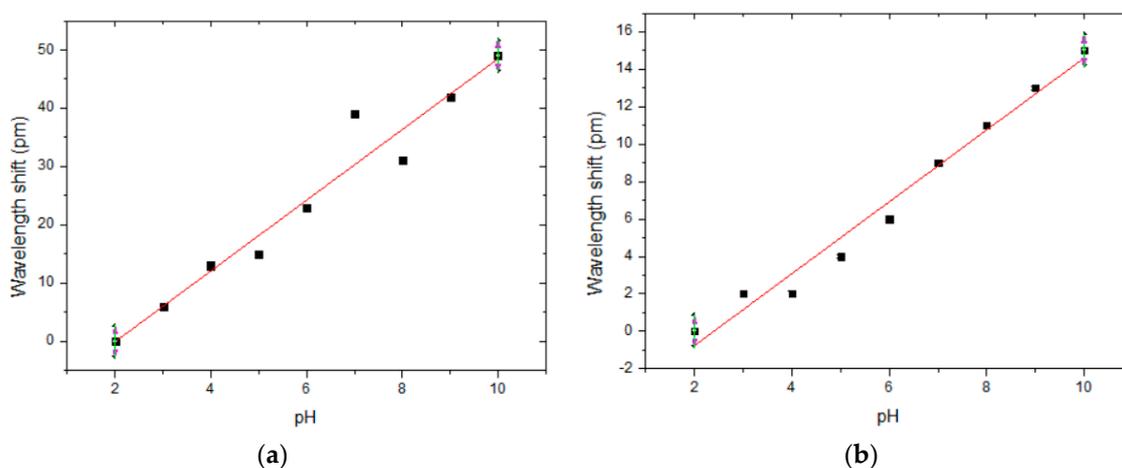


Figure 2. pH calibration curve for hydrogel coating on fibre sensors with (a) GO evaporated; (b) GO co-electroplated, techniques.

#### 4. Conclusions

A uniform coating of hydrogel across the sensing element was achieved through the increase of the surface roughness on the optical fibre. The two methods exhibited advantages and disadvantages. The GO evaporated + hydrogel had higher sensitivity ( $6.1 \pm 0.5$ ) pm/pH than the GO co-electroplated + hydrogel, ( $1.9 \pm 0.1$ ) pm/pH. However, at the same time, the first system was easily removed, whereas the other one was strongly attached to the optic fibre due to the strong bonding between GO and the metallic copper to the silica.

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